

問題7 データは 25°Cすなわち298K

窒素の粘性係数 $M = 28.02\text{g}$ $\sigma = 3.681\text{\AA}$ $\varepsilon/k = 91.5$ $kT/\varepsilon = 3.26$ $\Omega = 1.017$

$$\mu_{N_2} = 2.6693 \times 10^{-5} \frac{\sqrt{28.02 \times 298}}{3.681^2 \times 1.017} = 1.77 \times 10^{-4} \quad \mu_{N_2} = 1.77 \times 10^{-4} \frac{\text{g}}{\text{cm} \cdot \text{s}}$$

酸素の粘性係数 $M = 32.00\text{g}$ $\sigma = 3.433\text{\AA}$ $\varepsilon/k = 113\text{K}$ $kT/\varepsilon = 2.63$ $\Omega = 1.077$

$$\mu_{O_2} = 2.6693 \times 10^{-5} \frac{\sqrt{32.00 \times 298}}{3.433^2 \times 1.077} = 2.05 \times 10^{-4} \quad \mu_{O_2} = 2.05 \times 10^{-4} \frac{\text{g}}{\text{cm} \cdot \text{s}}$$

空気の粘性係数

Wilkeの半経験式
$$\mu_{air} = \frac{x_{N_2} \mu_{N_2}}{x_{N_2} + x_{O_2} \Phi_{NO}} + \frac{x_{O_2} \mu_{O_2}}{x_{N_2} \Phi_{ON} + x_{O_2}}$$

$$\Phi_{NO} = \frac{\left\{1 + (\mu_{N_2} / \mu_{O_2})^{1/2} (M_{O_2} / M_{N_2})^{1/4}\right\}^2}{\left\{8(1 + M_{N_2} / M_{O_2})\right\}^{1/2}} = \frac{\left\{1 + (1.77 / 2.05)^{1/2} (32.00 / 28.02)^{1/4}\right\}^2}{\left\{8(1 + 28.02 / 32.00)\right\}^{1/2}} = 0.992$$

$$\Phi_{ON} = \frac{\left\{1 + (\mu_{O_2} / \mu_{N_2})^{1/2} (M_{N_2} / M_{O_2})^{1/4}\right\}^2}{\left\{8(1 + M_{O_2} / M_{N_2})\right\}^{1/2}} = \frac{\left\{1 + (2.05 / 1.77)^{1/2} (28.02 / 32.00)^{1/4}\right\}^2}{\left\{8(1 + 32.00 / 28.02)\right\}^{1/2}} = 1.01$$

$$\mu_{air} = \left(\frac{0.8 \times 1.77}{0.8 + 0.2 \times 1.01} + \frac{0.2 \times 2.05}{0.8 \times 0.992 + 0.2} \right) \times 10^{-4} = 1.83 \times 10^{-4} \quad \mu_{air} = 1.83 \times 10^{-4} \frac{\text{g}}{\text{cm} \cdot \text{s}}$$

問題8

単原子分子の定積比熱は $C_V = \frac{3}{2} R$ 単位gあたり $C_V = \frac{3}{2} \frac{R}{M}$

定圧比熱は $C_P = \frac{5}{2} R$ 単位gあたり $C_P = \frac{5}{2} \frac{R}{M}$

Chapman-Enskogの理論では理想気体でないのだが、ここでは $C_P = C_V + R$ としよう

Euckenの式に代入して展開する。

$$\lambda = \left(C_P + \frac{5}{4} \frac{R}{M} \right) \times \mu_{SO_2} = \frac{15}{4} \frac{R}{M} \times \mu_{SO_2} = \frac{15 \cdot 1.987}{4M} \times 2.6693 \times 10^{-5} \frac{\sqrt{MT}}{\sigma^2 \Omega_\mu} \frac{\text{cal}}{\text{cm} \cdot \text{s} \cdot \text{K}}$$

式を整理して、さらに $\Omega_\mu = \Omega_k$ より

$$= 1.9890 \times 10^{-4} \frac{\sqrt{T/M}}{\sigma^2 \Omega_k} \frac{\text{cal}}{\text{cm} \cdot \text{s} \cdot \text{K}}$$